87075JLT Customer No. 01333

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of:

Sharon M. Simpson

PHOTOTHERMOGRAPHIC MATERIALS WITH IMPROVED NATURAL AGE KEEPING

Serial No. 10/826,780

Filed 16 April 2004

Group Art Unit: 1752

Examiner: LETSCHER, Geraldine

I hereby certify that this correspondence is being deposited today with the United States Postal Service as first class mail in an envelope addressed to Commissioner For Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

Sherryl A. Payne

Sure 13, 2005

Commissioner for Patents P.O. Box 1450 Alexandria, VA. 22313-1450

Sir:

DECLARATION UNDER 37 C.F.R. 1.131

- I, Sharon M. Simpson hereby say and declare that:
- (1) I am the inventor of the invention described and claimed in the present patent application identified above.
- (2) Since January 1999 I have been employed by Eastman Kodak Company at its facility located in Oakdale, Minnesota.
- (3) During my employment with Eastman Kodak Company, I have have been and am currently involved in the research and development of high-speed photothermographic materials, components used therein, and methods of imaging these materials to provide visible images.
- (4) The presently claimed invention was conceived and reduced to practice in the United States of America prior to November 17, 2003, the priority date of copending Application No. 10/715,199 that has been cited as prior art in the recent Office Action received from the USPTO.

- (5) Prior to November 17, 2003, I conceived of and reduced to practice:
- (I) An organic solvent-based photothermographic material comprising a support having thereon, one or more imaging layers comprising a hydrophobic binder and:
 - a. a photosensitive silver halide,
- b. in reactive association with the photosensitive silver halide, a non-photosensitive source of reducible silver ions comprising a silver carboxylate,
- c. a reducing agent for the reducible silver ions comprising a phenolic developer,
- d. an aliphatic or non-aromatic carbocyclic polycarboxylic acid that is present in an amount of from about 0.0004 to about 0.01 mol/mol of total silver (or from about 0.0015 to about 0.0375 g/m²), and
 - e. optionally, an X-radiation-sensitive phosphor,
- (II) An organic solvent-based X-radiation sensitive photothermographic material that comprises a support having on one side thereof, a photothermographic imaging layer comprising a hydrophobic binder and in reactive association:
- a. a photosensitive silver bromide or silver iodide, or mixture thereof, that has been chemically sensitized with a sulfur-containing chemical sensitizing compound, a tellurium-containing chemical sensitizing compound, or a gold(III)-containing chemical sensitizing compound, or mixtures of any of these chemical sensitizing agents,
- b. in reactive association with the photosensitive silver halide, a non-photosensitive source of reducible silver ions that comprises silver behenate,
- c. a reducing agent for the reducible silver ions that comprises a hindered phenol,
- d. one or more X-radiation-sensitive phosphors that are present in a total amount of from about 0.1 to about 20 mole per mole of total silver, the amount of total silver being from about 0.01 to about 0.05 mol/m², and

- e. one or more of citric acid, tartaric acid, maleic acid, fumaric acid, citraconic acid, mesaconic acid, tricarballylic acid, malonic acid, 1,2,3,4-butanetetracarboxylic acid, 1,2,3,4-cyclopentanetetracarboxylic acid, 1,3,5-cyclohexanetricarboxylic acid, and 1,2-cyclohexanedicarboxylic acid in an amount of from about 0.001 to about 0.004 mol/mol of total silver (or from about 0.004 to about 0.09 g/m²), and
 - (III) A method for forming a visible image comprising:
- A) imagewise exposing any of the photothermographic materials of the present invention to radiation to form a latent image, and
- B) simultaneously or sequentially, heating the exposed photothermographic material to develop the latent image into a visible image.
- (6) Exhibit A is a true electrophotographic copy of notebook pages 181, 182, and 184 in notebook CC0251 assigned to Sharon M. Simpson that are dated prior to November 17, 2003 and that describe a photothermographic composition and material as claimed in the present application, except that irrelevant information has been obscured.
- (7) In particular, notebook page 181 of Exhibit A describes a photothermographic composition as having a binder identified as "B79" a homogenate containing a photosensitive silver halide and non-photosensitive source of reducible silver ions identified as "CZ5XX-S 2321 Homogenate" and a reducing agent for the reducible silver ions identified as "Permanox." Notebook page 182 describes citric acid, a polycarboxylic acid identified as CA. Notebook page 184 describes the sensitometry of this sample after coating, drying, and imaging. These pages represent Example 1 of the patent application.

These features have been highlighted in yellow on pages 181, 182, and 184 for the Examiner's convenience.

(8) Exhibit B is a true electrophotographic copy of notebook pages 177-179 in notebook CC0251 assigned to Sharon M. Simpson that are dated prior to November 17, 2003 and that describe a photothermographic composition and material as claimed in the present application, except that irrelevant information has been obscured.

photothermographic composition as having a binder identified as "B79" a homogenate containing a photosensitive silver halide and non-photosensitive source of reducible silver ions identified as "CZ5XX-S 2303 Homogenate" and a reducing agent for the reducible silver ions identified as "Permanox." Notebook page 178 describes citric acid, a polycarboxylic acid identified as CA and the optional X-radiation-sensitive phosphor identified as YSrTaO₄. Notebook page 179 describes the sensitometry of this sample after coating, drying, and imaging. These pages represent Example 3 of the patent application.

These features have been highlighted in yellow on pages 177-179 for the Examiner's convenience.

- (10) Thus, Exhibits A and B demonstrate that conception and reduction to practice of the presently claimed invention were made prior to November 17, 2003.
- and all statements made on information and belief are believed to be true, and that these statements are made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

/9/os____

Date

Sharon M. Singson

Sharon M. Simpson

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	04,
Experiment Name:	Dental

To compare ZaV sensitization using MEOH or MEK/MEOH solvent and mix time

of solution before addition to Ag homogenate (2 min. vs, 30 min.) To compare B79 resin to B18/B16 resin effects on sensitization. To compare 2%I emulsion at 0.20um grain size (2335) with standard 6% I (2081 emulsion). Fisher solvent was used for all solution 0.14um emulsion soap 2321 made on

adds. Citric acid was added to the Ag layer without phopshor to follow RSK.

APR7_xts

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Red 1A Filter		Α	60C		B	60C	C ·	eoc	D		E	_ r		П	F	T	ľ.	
CZSXX - S	7-	2321	0.49/10		2321	0.4g/m	2321	0.49m		0.4g/m	2335		Mix Time	Temo	R.P.N		 	
Homogenate	1	S/Au	44C		S/Au	44C	S/Au	44C	S/Au	44C	S/Au		min.	F	 	New Lat of ME	K Imm Enter	 :
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	1	Fisher			Fisher		Floher		Fleher	<u></u>	Fleher	1	1	<u> </u>	L	L	1	1
	1	23622		<u> </u>]	I									<u> </u>	1
CC74 Lot ZaV-7		A			B		С		c		c	† ··· · · · · · · · · · · · · · · · · ·		1		A	 	
331g/m		8.2mf		1	8.2ml		8.2ml		8.2ml	-	8.2ml		40	!		0.0508g KSCX	-	
	Solution	Mix 2 mir			Mix 30mi		Max 30cm	<u> </u>	Miles 30mil		Max 30res			 	-			<u> </u>
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		2.59x	ļ	├	 			ļ			1	<u> </u>		<u> </u>		С	<u> 1</u>	
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			1										1			0.0508g KSC0	74/8.640 MEC	ЭН
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ZnBr2 225.19g/m	 	1.50m/	0.1699	 	1.50ml		1.50ml		1.50ml		1.50ml		+	 		40-1-200	 	
01012 223.13Y10	+	1.5011			1		1.30EIR		1.5Um		1.500		30			1.014g ZnBr2		
	ļ	ļ	1.190									L	<u> </u>			7.14g MEOH		
	+				 	ļ	<u> </u>		L		L						L	L
HP		2.00ml	0.20g		2.00ml	L	2.00ml		2.00ml		2.00ml		60			1.20gPHP/9.4	8g	
MeOH		i	1.58g										1			MeOH	Ī	
	T										 	 	+		-	STD Lot Today		
	1		 		_				 		+	 	 			10.0 COT 1008	<u>'</u>	
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		L		<u></u>														
Au(III)IpyrCI(CI)2		4.8ml			4.8ml		4.8ml		4.8ml		4.Bml		60			0.00520	lot G	
591g/m											-		 	\vdash		50.0g MeOH		
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														\sqcup			<u> </u>	<u></u>
288A		1.42g		<u> </u>	1.420		1.42g		1.42g		1.42g	<u> </u>	15				I	
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SM1881,16									12g/8g								L	L
													!					l
	1		Α	4.Or												A		
SSP			2.14g													12.84g BSP		
ÆK			24.20								-				_	145.29 MEK		
	 	26.349	2-20		26.34g		26.340	-	20 24-							14029 MEN		
		20.349			20-349		W-140		26.349		25.34g		10					
	_												<u> </u>					
HOH	1	pass	0.63g		P8:53		pass		pass		DR26					5.04gTHDV		
ÆK			1.50g													12.00gMEK		
	\Box												1	\vdash				
HZ 130.14g/m		pase	1.00g		DE:SS		pess		pess		0835		 	\vdash	-	0.00- 0-0-		
	i 	-			وديم		F10223		pess		pess			$\vdash \dashv$		8.00g PHZ/		ļ
EX			5.00g										1	$oxed{oxed}$		40.0g MEK		
otel Wt.																		
	L_ 7																	
HD4+PHZ]	8.13g			8.13g		8.13g		8.13g		8.13g		15	-		В		
HDHPHZ											 		 		_		abelan.	
											1		 		\dashv	12.789 THO! S		
	 										 					+ 36.0g PHZ		
	 												<u> </u>					
CPA+4-MPA		6.8g			6.8g		6.8g		6.8g		6.0g		15	1]	15.28g TCPA S	iolution+	
okation																28.93g 4-MPA		
CPA			0.35g						-							2.80gTCPA/		
EX	 		2.00g								 		1	-		16.00gMEK		
otal Wt.	 															16.UUDWEK		
AN W.	├ ──	Pees			pess		pees		pess		pees				1			
	LI										LI		1	ī	7			
MPA	L		0.45g										1			3.60g4MPA/		
юОН			0.50g										 			4.00g MeOH/		
EK			3.50g								 					28.0g MEK		
													 			Lu.uy MEN		
		i																
otal Wt.		pees		——	pees		pese		pees		page							

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Exhibit A

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	<u> </u>				2011	7		
	Coating							
Code	Ag	Gap Ag	TC Gap alone	TC+Ag Gap			AT 400FHC	214
					ATC4		AT 190F/16r	ev.
	A	3.8	2.2		ATC1		 	
2	A	3.8	2.2		AIOI			
	В	3.8	. 2.2	6	ATC1			-
	В	3.8	2		ATC			
	С	3.8	2		ATC	250 Ac + 0 5	iml CA Mix 5 n	nin.
	С	3.8			ATC :	25g Ag + 0.3	Bml CA Mix 5 n	nin.
	C C	3.8			ATC1		L	
	C	6.5			ATC1	25g Ag + 18	.2g YSrTaO4	Mix 5 min.
		1						
					ATO			
	D	3.8			ATC1	25g Ag ± 18	.2g YSrTaO4	Mix 5 min.
11	D	6.5	2.2	6.7	14101	25979 + 10		
10	E	3.8	2.2	6	ATC1			
	E	6.5			ATC1	25g Ag + 18	.2g YSrTaO4	Mix 5 min.
Batch					 		+	
1451/	1000-	 			 	0.035a Citr	ic Acid/3.5g N	EOH
MEK A21	1836g 5.758g		 		 			
CAB171-15S						1.50g BSP/	28.41g MEK	
J. 12 . 7 . 100							1704	
Total Wt.	2000g				1.70		ATC1 210g	
		ATC	010.5		BSP Soli	ution	21.0ml	
Batch		418g	313.5g 2.51g		DSP 3011	utton		
BTZ P382(72.0%)	-	3.34g 0.882g	0.662g		+			
CY27	4	0.242g	0.182g					
- · - ·								+
	1	Quart Jar			l			

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The foregoing disclosed to me on

Witness July

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RESEARCH / DEVELOPMENT Contifer By 183 000251 184 lotebook No. ate__ roblem CP 15226-6/00 Signature _ The foregoing disclosed to me on.

RESEARCH / DEVELOPMENT NOTEDOOK NOTE TO 251

EASTMAN KODAK COMPANY

Experiment Name:

Dental

To compare substituted diphenyl sulfides (ZaV-8 and ZaV-10) with the control ZaV-7. To compare the effects of MeOH or MEOH/MEK solvent for solution make of the ZaV compounds.

To study the effects of citric acid added to the silver layer containing phopshor for initial sensi and RSK.

The mix time of the YSrTAO4 was 5 minutes. The soap 2303 was used as made on contyaining

2081 emulsion (0.14um grains). RSK was followed.

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							10	60C	D		Ē						
Red 1A Filter			50C		В	60C 0.4g/m	C	0.4g/m		0.4g/m	2303				R.P.M.		
CZSXX - S			0.4g/m		S/Au	44C	S/Au	44C	S/Au	44C	S/Au		min.	(F)		New Lot of MEX	trom Fisher
Homogenete			44C		193.8g		193.80		193.8g	6%	193.80	6%					
		193.8g 23.90%	6%	67F	23.90%		23.90%	67F	23.90%	67F	23.90%	67F%	15	-	500		
	-14.12g			-14.120		-12.52g		-11.369	0.0g	-12.520	0.09			-	-		
MEK	-14,120	Fleher			Flaher		Fisher		Flaher	L	Fisher	ļ	<u> </u>		-		
		23622								L	<u> </u>		30	-	\vdash	A	
CC74 Lot ZaV-7		A 22.00		0.0608g 2	aV-7/8.64	gMEOH		J	<u> </u>	↓	00000	79/1.98gMEK		1	 	0.0500g KSCC	74/
331g/m	+	8.0ml			8.0ml			3/PgMEOI	1	-	S.Deni	/ 9 1.300	1	1		4.320 Fisher Mi	304.329 MEC
3319111						<u> </u>	6.0ml		5.0ml	 -	1.9x						
		2.531				ļ	1.9x	├		 	 						
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				 	├ ──	 		+	 	1					L_		<u> </u>
		<u> </u>		├	1.50ml	┼──	1.50mi	+	1.50ml		1.50ml		30	1	ļ	1.014g ZnBr2	
ZnBr2 225.19g/m		1.50ml	0.1699		11.3Unit	+	1,200.0	1	1			L	<u> </u>	 —	-	7.14g MEOH	
		 	1.190	-	—	 	+	1	\Box			 	 	+-	\vdash	1.20gPHP/9.48	
	+-	2.00ml	0.200	 -	2.00ml	1	2.00ml		2.00ml		2.00ml	 	6	+-	+	MeOH	•
PHP		2.00#1#	1.58g	1							┼	 		+	+-	STD Lot Today	<i>i</i>
MeOH	+-	1	·	1				1	4	 	+		+	+	+ -	T	
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Cool		cool			cool		cool		COOI	+	-	1					
			L	_	-		4.8ml	 	4.8ml	+	4.Bmi		6	0	1_	0.00629	lot G
Au(III) IDYTCI(CI)2		4.8ml	 		4.8ml	+	4.000	+	V.C	+					↓_	50.0g MeOH	
591g/m				┼	+	+	 		+			\perp		┷	4-		
				+	1.42g		1.420		1.420		1.420		 	5			+
C88A		1.429	├	┼	1						1			5 5	0	 	+
		cool	+	+	cool		cool	T	cool		cool			-	-	 	
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B79		1						→			+		 	1		IA	<u> </u>
			A	4.0x						+	+					14.98g BSP	↓
BSP			2 149								+				┷	169.49 MEK	
MEK			24.29		26.34g		26.349		26.349		25.349			10	-		+
		26.349	↓		20.349	+-	1									5.04gTHDY	+
		-	10.000	+	pass	_	pass		pess		pess				+	12.00gMEK	+
ТНОІ	_	pass	0.63g	+	-						-		+	+-		12.003	+
MEK		+	1.20								-		+	+-	+	8.00g PHZ/	
PHZ 130.14g/m	-+-	pass	1.000		pass		раяз		pess.	_	pess			1		40.0g MBK	
MEX	\dashv	1	5.000								+-	-					
Total WL		\perp			-		+-				+			\perp			
					-		8.13g	+-	8.139	+	8.13g			15	-	B	
THDMPHZ		8.13g	+-		8.13g	+-	0.130									12.78g THD + 36.0g PH	7
THDHPHZ	-		+		+-		_									- Jane Pro	+
											4			15	+	15.28g TCP	A Solution+
4 2004		6.89	+		6.89		6.89		6.8g		6.8g		+-	 -		28.93g 4-M	PA
TCPA + 4-MPA Solution	-+-	- June	+											_	-	2.80gTCPA	
TCPA			0.350										+	1		16.00gMBK	
MEX			2.00g			_			pess		pass	+-					
Total WL		pase			pess		pess		-		1			\Box			.+
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4-MPA			0.450					_						+		4.00g MeOl 28.0g MEK	
МеОН	_		0.500		\dashv		-	_						-+	-+	20.00 McA	-
MEK		pess	3.500		pess	_	pees		pess		pess			-+	+	+	
Total Wt.		_ pees			\neg						10.60			15			

KP 15226-6/00

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The foregoing disclosed to me on_

Exhibit B

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PAGE

EASTMAN KODAK COMPANY

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Coated 4/1/03 and proc	canad 3 da	Mi Mar			1		I		 _		T i							_			
2303/ 75g Gelatin .14us	m 6% AgX	0.40g PMT	Anole AoX	9.16mole?#	AgX AgX a	dd at 60C a	23.9% sol	ide													
S:Au+3 1:0.015 Lot H A	u for 454 S	<u> </u>																			
Au+3 added after Cool	and before	CBBA add																			
Zn Br2 before PNP			L		L	L		<u> </u>	<u> </u>	!											
1.0x P-382 used				└ ──	L	<u> </u>	<u> </u>		 -		 										
Kumar batch used at 2		<u> </u>		├ ──		<u> </u>					 										
BSP in TC studied at 2. Toner Batch Method/ Pr			<u> </u>		 	 			├──		 										
BSP added as a 4.00x								\vdash	i –	 	 									i	
B79 only			1			 				<u> </u>	 										
3040-03-03 YS/TaO4 4	um 18.20/	25.0g Ag		<u> </u>																	
Dry at 190F																					
STD, lot of PHP used		<u> </u>			ļ			<u> </u>	<u> </u>	<u> </u>	<u> </u>		<u> </u>								
40 min mix of ZaV use		ا		 		ļ.——				<u> </u>	};										
Fisher Solvent used to Methanoi Solvent used			-		V 700	├		<u> </u>	 		 		\vdash								
Cirtic Acid at 0.5ml to			/V MENIAL	CAT WILL 24	19-/9				 		 		-							 i	
Densi 1		<u> </u>	 	-						i	i i									i	
i li	ZaV						mil	AQ	Phosphor												
KeyID ZaV Type I				Cline Acid	BSP				a w		 						!		احصييا		
used		used	used	used	TC	Свр	Gap	Q/m2	g/m2		DMIN	D-HI	SPD#2 +4	AC-1	AC-S	AC-3	TC-1	TC-2	SPD#1 +4	SPD#3 -4	OMAX
12-11-00		MEOH	-			3.8		2.25	 	 	0.244	2.861	3.584	2.769	2.023		0.315	0.942	3.911	-+	2111
	1.58x	MEOH	 		<u> </u>	3.8			-		0.252	2.986		2.994	2.334		0.318				3.2
14/201					ř	1	<u>~</u>		 	 	1					7,547		4.47		i	
8 ZaV-93	1,34	MEOH	I		Y	3.8			1	i	0.271	3.15			3.261	0.773	0.3161		4.119		3.378
9 ZaV-93	1_9x	MEOH	18.20		у	6.5			76		0.797	4.588		3.742	4.506	5.326	0.2991	0.953	4.527	3.772	4.868
	1_9 ₂	MEOH	18.20	y		8.6	8.5		81		0.831	4.712		3,808	4.545	4.883	0.312		4,496	3.731	4,896
11 ZaV-93	1.02	МЕОН	18.20	 	 	6.5	8.5		76		0.85	4,747	4,199	3,797	4.346	5.317	0.31	0.941	4.557	3.7811	4.913
2 ZaV-7	2.53m	MEKMEO	<u></u>	-	 	3.8		2.26		 	0.254	2.984	3.724	3.276	2.326	0.604	0.295	0.991	4.03	2.291	3237
		MEKMEO		-	-	6.5			82		1690	4.509			4.458	4.591	0.385			3.571:	4,841
		1									:		1				i				
4 ZaV-7 12	2.53x	MEOH			y	3.8					0.264	2.986			2.334		0.297				3.261
		MEOH	18.2g		у	6.5			82		0.779	4,526		3.775	4.549	4.706				3.7841	4.865
		MEOH	18.20	Υ	<u> </u>	8.5	8.6		79		0.811!	4,589			4.534	3.93	0.251	0.909	4.518	3.772	4,859
7 ZaV-7	2534	MEOH	18.20		 	6.5	8.5		80	 	0.807	4.599	4.202	3.85	4.469	4.512	0.285	0.97	4.536	3.7781	4.509
13 ZaV-79	عار	MEKMEO	 			3.6	8	2.2	 	 	0.257	2,985	3,838	3.336	2.55	0.522	0.303	1.014	4,142	2,325	1,249
14 ZaV-79	90	MEKANEO			y	6.5	8.7		78		0.745	4.661	4.252	3.941	4.536	4.867		0.967	4.5781	3.8471	4.876
ZaV-93 -carbonyl pheny									1		ļ										
ZaV-79 - carbonyl diphe	nyl substitu	pent = ZaV-1	10		ļ			<u> </u>	<u> </u>	-	:										——
3 month aging								-		-	! .						i				\dashv
1 ZaV-93	58x	MEOH		\vdash	· -	3.8	6	2.25	i 	 	0.295	2.81	3,663	2.292	1,795	-	0.365	0.831	4.075	-	3.14
		MEOH		\vdash	у	3.8	6				0.28	2,911		2.461	2.088	0.501	0.352	0.853	4.043	2.11;	3.193
									1											i	
		MEOH			<u> </u>	3.8	- 6	2.18		<u> </u>	0.304	3.025	3.889		2.701	0.438	0.339	0.8721			3.207
			18.20	<u> </u>	ly	6.5	8,7		79		1,111:	4.665		3.468	4.138	3.8721	0.33	0.863		3.816	4.867
			18.2g	Y		6.5 8.5			76		0.962	4.659		3.4571	4.273 4.185	3,359	0.337	0.824			4.86
111247-53	.34	MECA	18.20				8.3		 	 	1.18	4.653	1 4212	3.43	103	1	- 0-002	0.7-0.	3.7		
2 ZaV-7 :2		MEXMEO			Υ	3.8		2.26			0.3	2914	3.784	2.648	2.07		0.368	0.832			3.265
31ZaV-7 12	2.534	MEKMEO	18.2g		y	6.5	8.7		82	1	0.906	4.454	4.171	3.279	3.757	3.345	0.307	0.866	4.58	3.648:	4.801
		L	-	<u> </u>		<u> </u>					1							0.000		3.00	
		MEOH	18.20		<u>y</u>	3.8	8.7		82	 	0.31	2.863		2.649 3.274	1.875		0.341	0.869 0.747			3.202 4.536
			18.2g	L	<u>r</u>	8.6	8.6		79		1.034	+ 4.483 + 4.22			3.621		0.31	0.747	4.777	3.676	4.846
			18.20			6.5			80		1,1481	4.485			3.832		0.683			3.674	4.85
									1												
		MEKAMED			у	3.8	. 6				0.284	2.913			2.263				4.23	2.292	3.218
14 ZaV-79 11	202	MEKAMEO	18.20		Υ	8.5	8.7		78		0.955	4.575	4.257	3.552	4.099	3.97	0.299	0.914	4.848	3.799	4.839
Chance in Same				-				-	 	 	 		\vdash				i				
Change in Sensi	.58a	МЕОН			·	3.8	6	225	 	 	0.051	-0.051	0.099	-0.477	0.220	#VALUE!	0.05	-0.111	0.164	WALLE	0.029
		MEOH			y	3.8					0.028	-0.075		-0.533	-0.246		0.034		0.153		-0.007
		MEOH			γ.	3.8	- 6	2.18			0.033;	0.125	0.092	-0.6491	-0.58	-0.335	0.023				-0.169
			18.20		y	6.5	8.7		79		0.314	0.023	0.106	0.274	-0.368	-1.454	0.0011	-0.09	0.147	0.0441	0.007
			18.20	у		8.5			76		0.131:	-0.103 -0.088	0.128	-0.351 -0.347	-0.272 -0.161	-0.958 -1.958	0.025	-0.12 -0.196		0.0831	0.053
11 ZaV-93 1			18.20				ده			+	0.331	-0.000	V.113	7,-1	3.101	-7,556	3.432	3,130	7,7-5	1	
2 ZaV-7 2	2.53%	MEKMEO	H		у	3.8	6	2.26	i -		0.046	-0.07	80.0	-0.626	-0.258	0.002	0.073	0.159	0.138	-0.015	0.028
		MEKMEO			у	8.5	8.7		82		0.215	-0.155	0.076		-0.701		-0.078	-0.013		-0.023	-0.04
										1										i	
		MEOH			DY	3.8	B				0.046	-0.103			-0.459		0.043	0.13		-0.097	0.059
			18.20	<u></u>	Υ	6.5	8.7		82		0.255	-0.143 -0.369	0.03	-0.501 -0.5771	-0.931 -0.913	-1.477 -1.38	0.089		0.11	-0.098	0.029
		MEOH	18.2g 18.2g	Y	 	6.5			80		0.085	-0.369 -0.114		-0.403	-0.913		0.398		0.152	-0.104	0.019
1/247-7			10.44		 	1	- 3.0			 	1	7.1.7	, , , , , , , , , , , , , , , , , , ,			2.000				9	¥~
		MEKMEO			Y	3.8					0.027	-0.072	0.029	-0.536	-0.267		0.035	-0.109		-0.033	0.033
		MEKMEO			Y	6.5	8.7		78	L	0.21	-0.086	0.015	-0.389	-0.537	-0.897	SVALUE!	-0.053	0.07	-0.048	-0.037

Signature

The foregoing disclosed to me on.